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Student exploration collision theory answers student exploration. Observe a chemical reaction with and without a catalyst. Determine the effects of concentration temperature surface area and catalysts on reaction rates. While the computers are coming up i encourage students to answer the prior knowledge questions.

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Student Exploration: Collision Theory Vocabulary: activated complex, catalyst, chemical reaction, concentration, enzyme, half-life, molecule, product, reactant, surface area Prior Knowledge Questions (Do these BEFORE using the Gizmo.) 1. Suppose you added a spoonful of sugar to hot water and another to ice-cold water. Which type of water will cause the sugar to dissolve more quickly?

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What is the collision theory? It is the idea that particles have to collide in order to react and they have to collide hard enough (with enough energy to break the bonds - activation energy). How does a higher temperature increase the rate of reaction? When the temperature is increased the particles have more energy and move quicker.

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### Student Exploration: Collision Theory

Student Exploration Collision Theory Gizmo Answers hardest task as they had to figure out what it was about the catalyst that was speeding the. Using your knowledge of collision theory. Collision Theory Gizmo Answer Key Reactants are substances that enter into a reaction, and products are substances produced by the reaction. The Collision Theory Page 11/26

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## Student Exploration Collision Theory Answers

Student Exploration: Plate Tectonics slowly-moving sections of Earth's crust. At plate boundaries, plates collide, move apart, move under or over each other, or slide past one another. The theory of plate tectonics describes how the plates move, interact, and change the physical landscape.

## Student Exploration Plate Tectonics Answer Key

6. Propose a theory: Based on what you have observed, explain why you think phase changes occur. If possible, discuss your theory with your classmates and teacher. Sample answer: Phase changes occur because of the energy of molecular motion. As heat is added to a solid, the molecules break out of their bonds and begin to move freely, causing

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