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solution of all  
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sure it is balanced as  
written; balance if  
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know Formula weights or  
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(abbreviated FW or MW) for  
relevant compounds: you will  
often need to calculate  
these for yourself using the  
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reactants and the products  
as well as the accurate  
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are Based on Chemical  
Formulas Let's learn some  
terms used in Stoichiometry

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first. Formula Mass: It is the sum of the atomic weights of the various atoms present in the molecule of the substance. For example, we can calculate the formula mass of  $\text{Na}_2\text{S}$  as  $2(23) + 1(32) = 78$



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Equation stoichiometry problems and shows how the skills developed in Section 10.1 can be mixed with the new skills developed in this section. Section 13.3 completes our process of describing equation

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*AACT*

Stoichiometry as the calculation of products and reactants in a chemical reaction. It is basically concerned with numbers. Stoichiometry is an

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important concept in chemistry that helps us use balanced chemical equations to calculate amounts of reactants and products. Here, we make use of ratios from the balanced equation.

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*What is Stoichiometry?*

*Balancing Equations,  
Stoichiometric ...*

The relating of one chemical substance to another using a balanced chemical reaction is called stoichiometry. Using stoichiometry is a

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fundamental skill in chemistry; it greatly broadens your ability to predict what will occur and, more importantly, how much is produced. Let us consider a more complicated example.

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