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advise on diving  
medicine issues,  
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essential to  
understanding diving  
medicine is the  
relationship between  
gas partial pressure  
and the ability of a  
gas to dissolve in a

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solute. Henry's Law states that the amount of gas which will dissolve in a liquid at any given temperature is a function of: 1) the partial pressure of the gas in contact with the liquid, and 2) the solubility coefficient of the gas in the liquid.

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Diving medicine, also called undersea and hyperbaric medicine (UHB), is the diagnosis, treatment and prevention of conditions caused by humans entering the

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*Diving Medicine: A  
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