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Part I deals with the formulation, and Part
11 with the methods of exact and
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Solution : The equation of the heat transfer conduction : $Q/t =$ the rate of the heat conduction, $k =$ thermal conductivity, $A =$ the cross-sectional area, $T_2 =$ high

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temperature, T_1 = low temperature, T_2 = high temperature, $\Delta T = T_1 - T_2$ = The change in temperature, l = length of metal.

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Solution: The equation of the heat transfer conduction : $Q/t =$ the rate of the heat conduction, $k =$ thermal conductivity, $A =$ the cross-sectional area, $T_2 =$ high temperature, $T_1 =$ low temperature, $T_1 - T_2 =$ The change in temperature, $l =$ length of metal. Both rods have the same size so that A eliminated from the

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Solution Manual The solutions contain 20% (200 mg/mL) or 10% (100 mg/mL) acetylcysteine, with disodium edetate in purified water. Sodium hydroxide and/or hydrochloric acid is added to adjust pH (range 6.0 to 7.5).

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- It can be used practically in heat transfer for a relatively short time and/or in a relatively thick material
 - The governing equation with no bulk flow and no heat generation is
 - The boundary conditions are
 - The initial condition is
- $$2 \times T \text{ t } T \text{ T } x \text{ 0 } T \text{ s } T \text{ x } T \text{ i } T \text{ t } 0 \text{ T } i$$

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Conduction Heat transfer: Unsteady state

Heat transfer occurs at a lower rate in materials of low thermal conductivity than in materials of high thermal conductivity. For instance, metals typically have high thermal conductivity and are very efficient at conducting heat, while the opposite is true for insulating materials like Styrofoam

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